Received 28 July 2006

Accepted 10 August 2006

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Yavuz Köysal,^a* Şamil Işık,^a Kemal Sancak^b and Yasemin Ünver^b

^aDepartment of Physics, Faculty of Arts and Sciences, Ondokuz Mayıs University, Kurupelit, 55139 Samsun, Turkey, and ^bDepartment of Chemistry, Faculty of Arts and Sciences, Karadeniz Technical University, Trabzon, Turkey

Correspondence e-mail: yavuzk@omu.edu.tr

Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.003 Å R factor = 0.042 wR factor = 0.127 Data-to-parameter ratio = 15.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

3,3'-Diisopropyl-4,4'-(hexane-1,6-diyl)bis-[1*H*-1,2,4-triazol-5(4*H*)-one] dihydrate

The triazole molecule of the title dihydrate, $C_{16}H_{28}N_6O_{2}$. 2H₂O, possesses a crystallographically imposed centre of symmetry and shows normal values of bond lengths and angles. The uncoordinated water molecules are involved in intermolecular N-H···O, O-H···O and O-H···N hydrogen bonds, which stabilize the crystal packing.

Comment

1,2,4-Triazole and its derivatives exhibit various pharmacological properties such as antimicrobial (Holla et al., 1998; Ersan et al., 1998), anti-imflammatory, analgesic, antitumoral, antihypertensive and antiviral (Kritsanida et al., 2002). In addition, compounds incorporating thidiazole and 1,2,4-triazole have been produced as antimicrobial agents (Holla et al., 1998; Ersan et al., 1998). Some of the azole derivatives used as common antibiotics, such as amphotericin B, exhibit toxic effects on humans along with antimicrobial effects (Collin et al., 2003). Although different antimicrobial agents are used in the treatment of microbial infections, an increasing resistance to these drugs is observed (Collin et al., 2003). Therefore, the search for and synthesis of new antibiotics different from commonly used ones is of current importance. We present here the crystal structure of the title compound, (I) (Fig. 1), a new 1,2,4-triazole derivative.



The triazole molecule possesses a crystallographically imposed centre of symmetry at the mid-point of the central C-C bond, and shows normal values of bond lengths and angles (Allen et al., 1987), cooresponding to those observed in 4,4'-(butane-1,4-diyl)bis[3-ethyl-1H-1,2,4-triazol-5(4H)-one] and 4-hydroxy-3-n-propyl-1H-1,2,4-triazol-5(4H)-one (Ocak

© 2006 International Union of Crystallography All rights reserved

organic papers

Ískeleli *et al.*, 2005). The uncoordinated water molecules are involved in intermolecular $N-H\cdots O$, $O-H\cdots O$ and $O-H\cdots N$ hydrogen bonds (Table 1), which stabilize the crystal packing (Fig. 2).

Experimental

N'-(1-Ethoxy-2,2-dimethylethylidene)hydrazine carboxylic acid ethyl ester, (II), was obtained from the reaction of N'-(1-ethoxy-2,2-dimethylethylidene)imino ester hydrochloride (0.01 mol) with ethyl carbazate (0.01 mol). 1,6-Bis(3-isopropyl-4,5-dihydro-1H-1,2,4-triazole-5-one-4-yl)hexane was obtained from the reaction of (II) and 1,6-diaminohexane. To a solution of (II) (0.01 mol) in water (50 ml), 1,6-diaminohexane (0.01 mol) was added. The mixture was refluxed for 4 h and the resulting precipitate was filtered off. The solid obtained was recrystallized from water (yield 51.01%; m.p. 506–507 K).

Z = 2

 $D_r = 1.196 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation

Prism, colourless

 $0.80 \times 0.52 \times 0.28 \; \text{mm}$

14522 measured reflections 2033 independent reflections

 $w = 1/[\sigma^2(F_0^2) + (0.077P)^2]$

where $P = (F_0^2 + 2F_c^2)/3$

+ 0.2194*P*]

 $\Delta \rho_{\text{max}} = 0.17 \text{ e} \text{ Å}^{-3}$

 $\Delta \rho_{\rm min} = -0.18 \text{ e} \text{ Å}^{-3}$

 $(\Delta/\sigma)_{\rm max} < 0.001$

1592 reflections with $I > 2\sigma(I)$

 $\mu = 0.09 \text{ mm}^{-1}$

T = 293 (2) K

 $R_{\rm int} = 0.045$

 $\theta_{\rm max} = 26.0^\circ$

Crystal data

 $\begin{array}{l} C_{16}H_{28}N_6O_2\cdot 2H_2O\\ M_r = 372.47\\ Monoclinic, P2_1/c\\ a = 7.7311 \ (6) \ \AA\\ b = 10.7183 \ (8) \ \AA\\ c = 12.691 \ (1) \ \AA\\ \beta = 100.416 \ (6)^\circ\\ V = 1034.30 \ (14) \ \AA^3 \end{array}$

Data collection

Stoe IPDS-2 diffractometer ω scans Absorption correction: integration (X-RED32; Stoe & Cie, 2002) $T_{\min} = 0.976, T_{\max} = 0.991$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.042$ $wR(F^2) = 0.127$ S = 0.922033 reflections 134 parameters H atoms treated by a mixture of independent and constrained refinement

Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
N4-H4···O33 ⁱ	0.86	1.89	2.7389 (18)	170
O33-H66···O1 ⁱⁱ	0.853 (16)	1.885 (16)	2.7372 (17)	176 (2)
O33-H77···N5	0.878 (16)	2.033 (17)	2.9105 (19)	177 (2)

Symmetry codes: (i) -x, -y, -z + 1; (ii) x + 1, y, z.

Water H atoms were located in a difference Fourier map and isotropically refined with O–H distance restraints of 0.85 (3) Å. The amino and C-bound H atoms were positioned geometrically (N–H = 0.86 Å and C–H = 0.93–0.97 Å) and refined using a riding model, with $U_{\rm iso}(\rm H) = 1.2U_{eq}$ or $1.5U_{eq}$ (parent atom).

Data collection: X-AREA (Stoe & Cie, 2002); cell refinement: X-AREA; data reduction: X-RED32 (Stoe & Cie, 2002); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s)



Figure 1

View of the title compound, showing the 50% probability displacement ellipsoids and the atom-numbering scheme [symmetry code: (i) -1 - x, -y, -z].



Figure 2

The packing, viewed down the b axis. Hydrogen bonds are shown as dashed lines.

used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPIII* (Johnson & Burnett, 1996); software used to prepare material for publication: *WinGX* (Farrugia, 1999) and *PARST* (Nardelli, 1995).

References

Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1–19.

Collin, X., Sauleau, A. & Coulon, J. (2003). J. Am. Chem. Soc. 97, 1354–1358. Ersan, S., Nacak, S. & Berkem, R. (1998). Il Farm., 53, 773–776.

- Farrugia, L. J. (1999). J. Appl. Cryst. **32**, 837–838.
- Holla, B. S., Poorjary, K. N., Rao, B. S. & Shivananda, M. K. (1998). *Eur. J.*
- Med. Chem. **37**, 511–517. Johnson, C. K. & Burnett, M. N. (1996). ORTEPIII. Report ORNL-6895. Oak Ridge National Laboratory, Tennessee, USA.
- Kritsanida, M., Mouroutsou, A., Marakos, P., Pouli, N., Papakonstantinou-Garoufalias, S., Pannecouque, C., Witvouw, M. & Clercq, E. (2002). *Il Farm.*, 57, 253–257.

Nardelli, M. (1995). J. Appl. Cryst. 28, 659.

Ocak Ískeleli, N., Işık, Ş., Sancak, K., Şaşmaz, S., Ünver, Y. & Er, Y. (2005). *Acta Cryst.* C61, 0363–0365. Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.

Stoe & Cie (2002). X-AREA (Version 1.18) and X-RED32 (Version 1.04). Stoe & Cie, Darmstadt, Germany.